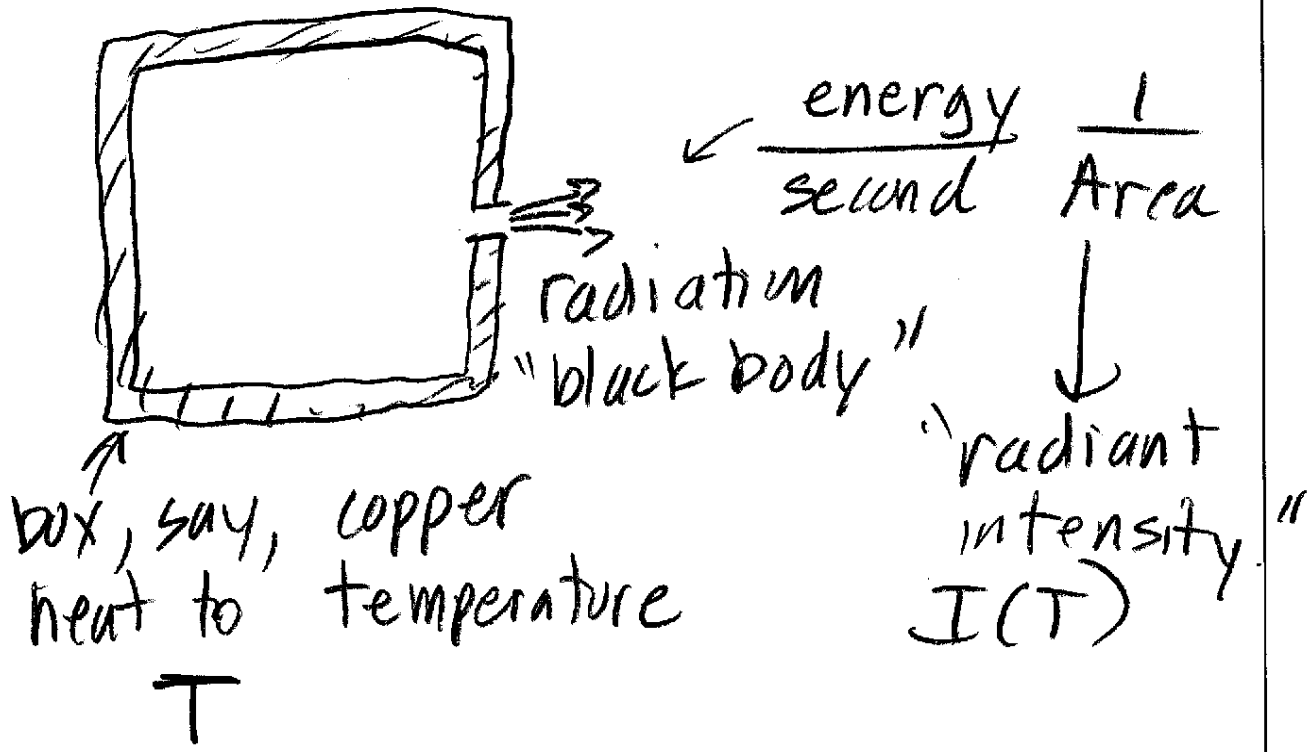


Hot objects emit light.

"red hot", "white hot"
even a "black body" emits radiation
when above $T > 0$

↑
Kelvin

empirical work was crucial



① $I(T) = \sigma T^4$

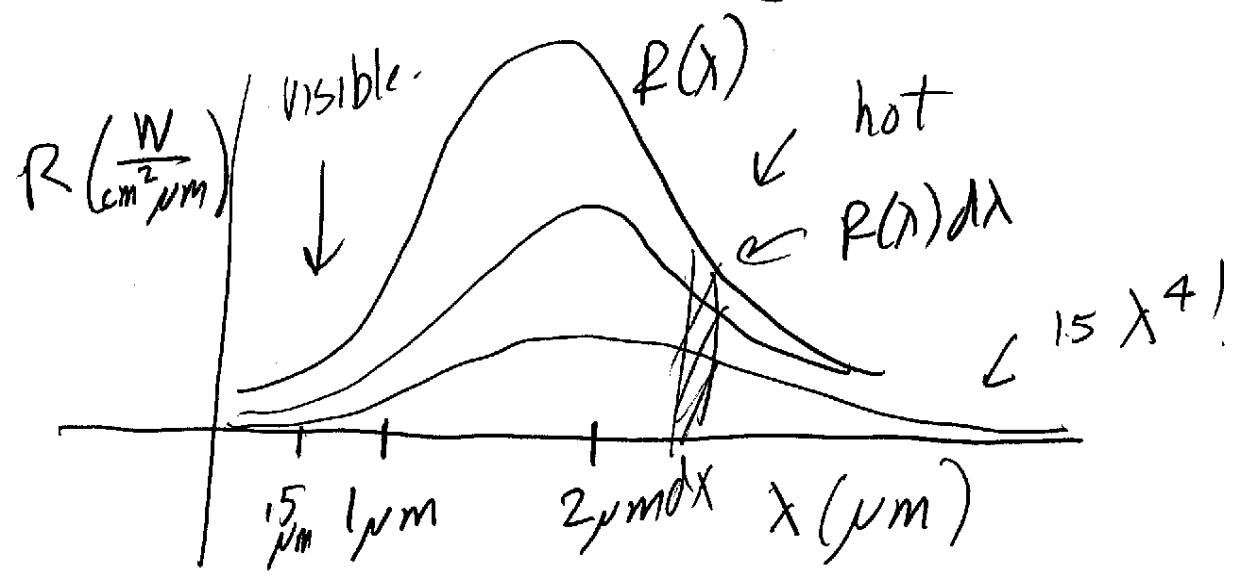
↑
Stefan-Boltzmann
 $= 5.67 \cdot 10^{-8} \frac{W}{m^2 K^4}$

→ explainable by deeper physics.

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② Spectral Radiance

put this radiant intensity in intervals of wavelength...



$$I(\lambda) = \sigma T^4 = \int_0^{\infty} R(\lambda) d\lambda$$

Form a mystery...

Rayleigh - Jeans'

$$R(\lambda) = \frac{2\pi^5 k T}{15 \lambda^4}$$

$\rightarrow \infty$ as $\lambda \rightarrow 0 !!$

$$(3) \lambda_{\max} T = 2898 \mu\text{m} \cdot \text{K}$$

$$\lambda_{\max} = \frac{2898 \mu\text{m} \cdot \text{K}}{T (\text{K})}$$

$$(i) \text{ Body } T = 98.6 \approx 37 \text{ C}$$

$$\approx 310 \text{ K}$$

$$\lambda_{\max} \approx 9.34 \mu\text{m}$$

$$(ii) \text{ Sun, } \lambda_{\max} \approx 500 \text{ nm} \approx \frac{1}{2} \mu\text{m}$$

$$T = \frac{2898 \mu\text{m} \cdot \text{K}}{1/2} \approx 5800 \text{ K}$$

Total Radiated power: ?

$$\text{body: } I(T) = \sigma T^4$$

$$\approx (5.67 \cdot 10^{-8}) (310)^4$$

$$\approx 524 \text{ W/m}^2$$

$$\approx 1.6 - 1.9 \text{ m}^2$$

Body
Area

$$\text{Energy Radiated} = 840 - 1000 \text{ W}$$

$$\text{Sun... } 4 \cdot 10^{26} \text{ W}$$

Planck's Radiation Law

Mathematical Fit to Empirical Data

$$R(\lambda) = \frac{a}{\lambda^5} \frac{1}{e^{\frac{b}{\lambda T}} - 1}$$

λ really long --

$$e^{\frac{b}{\lambda T}} \sim 1 + \frac{b}{\lambda T}$$

get $R(\lambda) \sim \frac{a}{\lambda^5} \frac{1}{\frac{b}{\lambda T}}$ Rayleigh

$$\approx \left(\frac{a}{b}\right) \frac{T}{\lambda^4} \approx \frac{2\pi^5 c^5 k T}{15 h^3 \lambda^4}$$

$$\frac{a}{b} = 2\pi^5 c k$$

λ really small.

Derivation: atoms oscillators but can only have energies

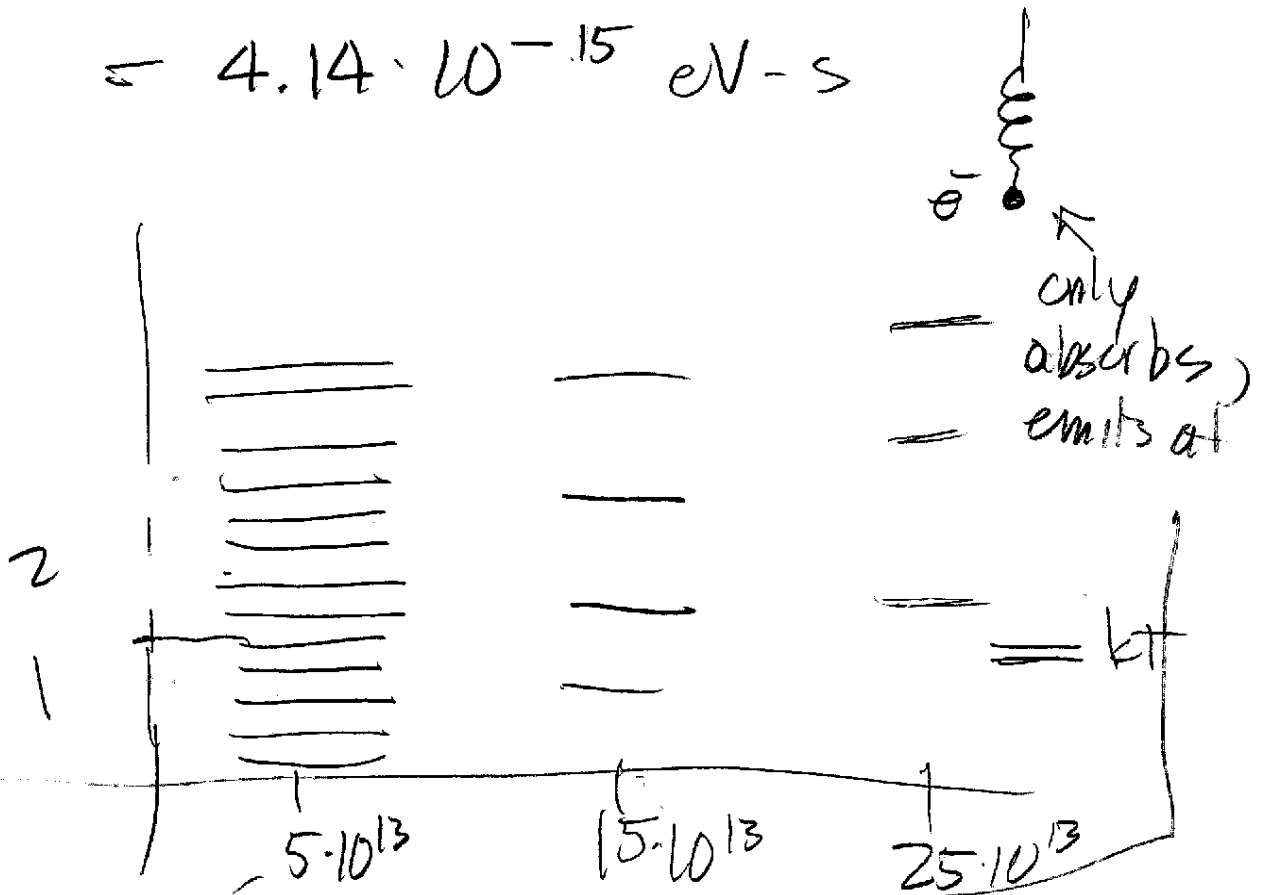
$$E = n h \nu = \frac{n h c}{\lambda}$$

$$R(\lambda) = \frac{2\pi^5 c^2 h}{15 \lambda^5} \frac{1}{e^{\frac{hc}{\lambda k T}} - 1} \quad \frac{2\pi^5 c^5 h}{15 k^4} = 2\pi^5 c k$$

numerical value of h

$$h = 6.63 \cdot 10^{-34} \text{ J} \cdot \text{s}$$

$$= 4.14 \cdot 10^{-15} \text{ eV} \cdot \text{s}$$



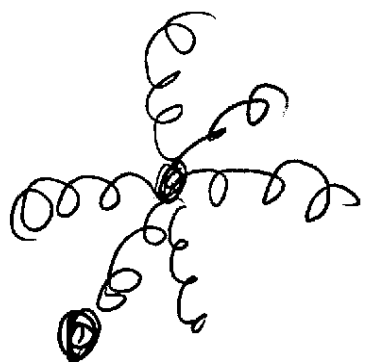
$$e^{-\frac{hc}{\lambda kT}} = e^{-\frac{h\nu}{kT}} \leftarrow \text{Boltzmann factor}$$

$$h\nu = 4.14 \cdot 10^{-15} \cdot 5 \cdot 10^{13}$$

$$\approx 20 \cdot 10$$

$$R(\lambda) \sim e^{-\frac{h\nu}{kT}} \quad \text{time}$$

Heat Capacity of Solids



$$E_{int} = 6 \times \left(\frac{1}{2} kT\right) N_A$$

$$= 3RT$$

↑
vibrations
 $\gamma \rightarrow$ spring constant.

crystal prop
↓
quantized, $h\nu_c = kT_E$

$$E_{int} = \frac{h\nu}{e^{\frac{h\nu}{kT}} - 1} \times 3N_A$$

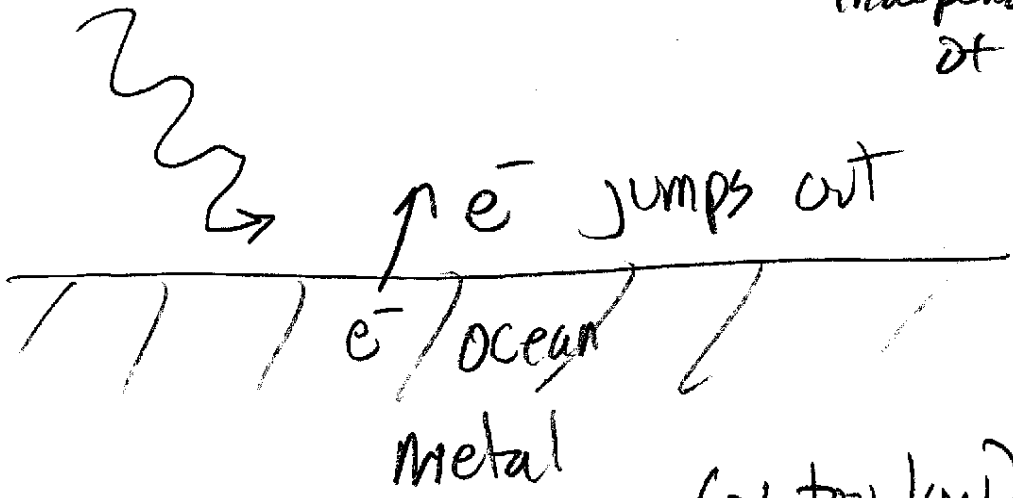
$$= 3RT_E \frac{1}{e^{\frac{T_E}{T}} - 1}$$

$$C_V = \frac{dE_{int}}{dT} = 3RT_E \frac{\frac{T_E}{T^2}}{\left(e^{\frac{T_E}{T}} - 1\right)^2}$$

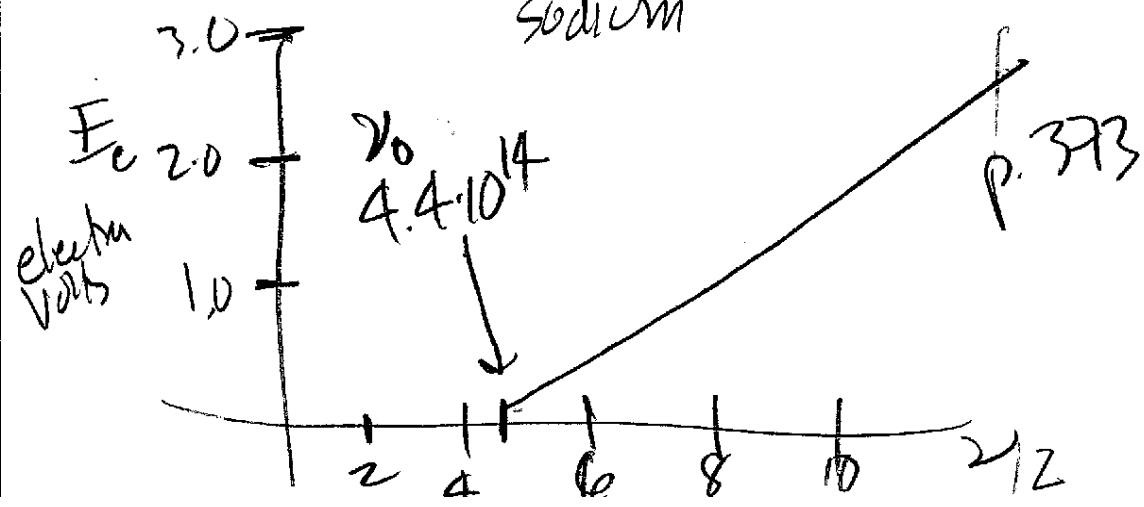
$$= 3R \left(\frac{T_E}{T}\right)^2 \left(\frac{1}{e^{\frac{T_E}{T}} - 1}\right)^2$$

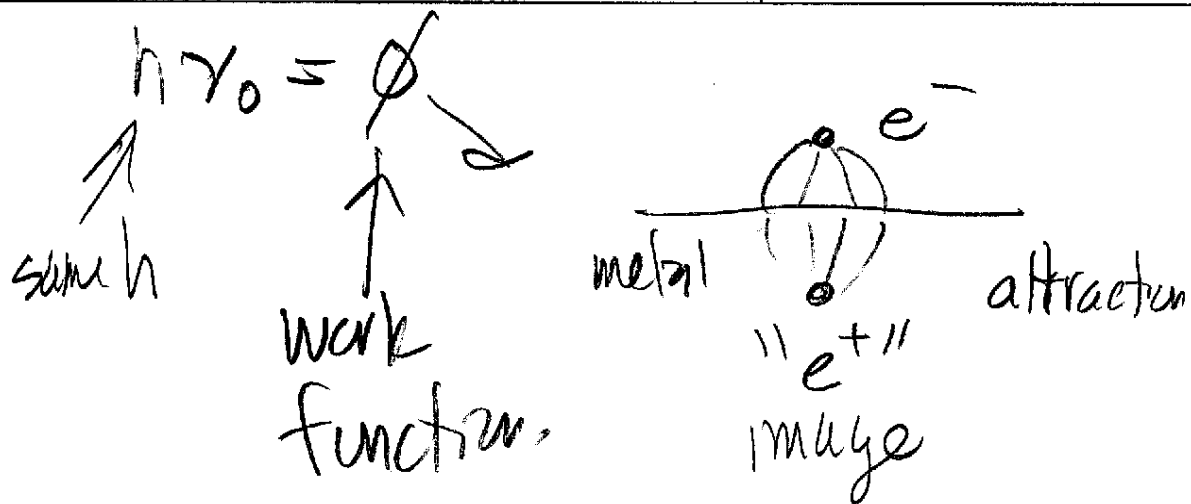
Photo electric Effect (Einstein's Nobel...)

$\lambda, \nu = \frac{c}{\lambda}$ Classically: $\theta = \frac{1}{2}(E^2 + B^2)$
 independent of λ, ν



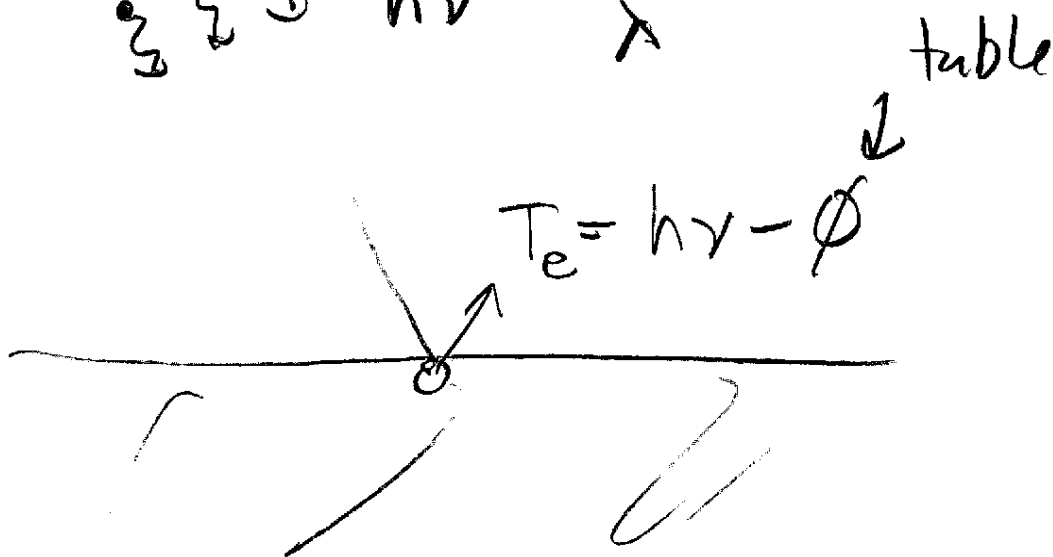
- ① if λ too long, no amount of intensity makes it happen
 - ② ν above ν_0 , ($\lambda < \lambda_0 = \frac{c}{\nu_0}$), get electrons.
 - ③ No time delay
 - ④ must feeble.
- sodium





Pictur

each energy
 $h\nu = \frac{hc}{\lambda}$

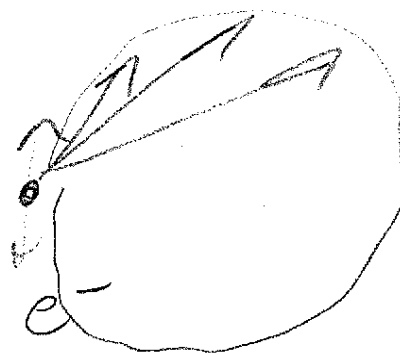
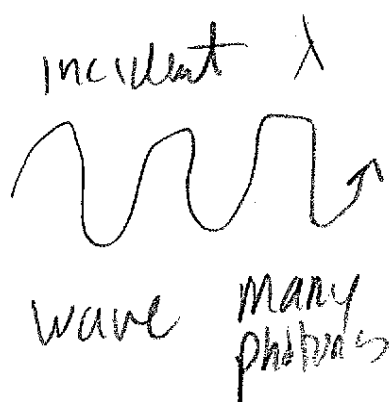


Compton Effect

x-rays behave like particles when they scatter of electrons.

Older - Thomson Scattering

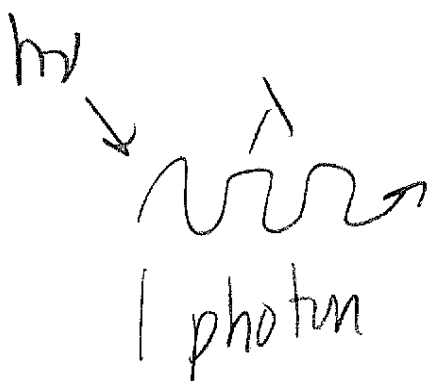
Thomson Scattering



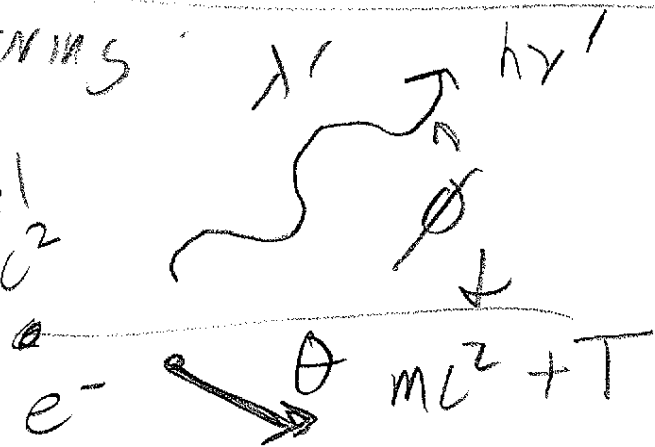
radiation
same frequency
 λ
many photons

ELASTIC
free $\dots \pi$
out of phase

Compton Scattering



initial
 mc^2



LOOKS INELASTIC
(actually, wrong frame)

$$h\nu + mc^2 = h\nu' + E_e$$

$$E_e = \gamma mc^2$$

$$E_e = mc^2 + T$$

$$\frac{hc}{\lambda} = \frac{hc}{\lambda'} + mc^2(\gamma - 1)$$

NO angles!

$$\gamma = \frac{1}{\sqrt{1 - \beta^2}}$$

Concept: momentum of photons

At rest: $E_e = m_e c^2$

Momentum $|\vec{p}_e|$: $E_e = \sqrt{(m_e c^2)^2 + (c|\vec{p}_e|)^2}$
 $\approx (m_e c^2) + \frac{|\vec{p}_e|^2}{2m_e}$

↑ no c

$$E_e^2 - (c|\vec{p}_e|)^2 = m_e^2 c^4$$

→ photon,

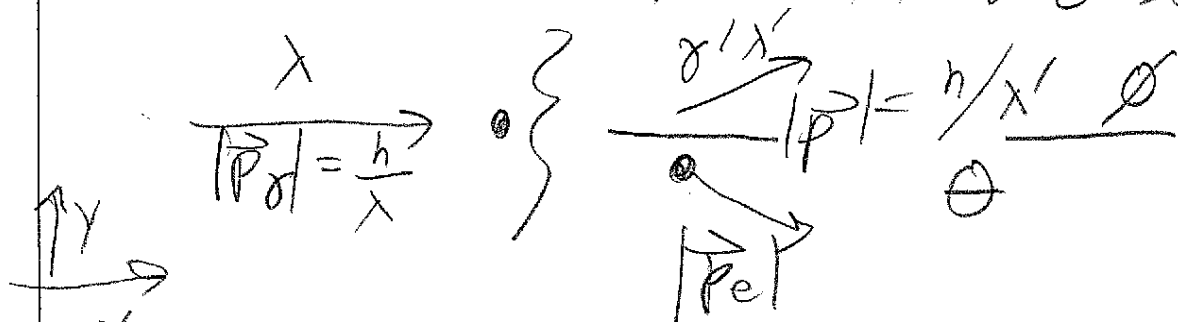
$$E_\gamma^2 - (c|\vec{p}_\gamma|)^2 = m_\gamma^2 c^4 = 0!$$

$$E_\gamma = c|\vec{p}_\gamma|$$

$$h\nu = c|\vec{p}_\gamma|$$

$$|\vec{p}_\gamma| = \frac{h\nu}{c} = \frac{h}{\lambda}$$

Momentum conservation in γ - e^- scattering



$$\frac{h}{\lambda} = \frac{h}{\lambda'} \cos \theta + |\vec{p}_e| \cos \theta$$

$$y: \quad 0 = \frac{h}{\lambda'} \sin \phi - |\vec{p}_e| \sin \phi$$

$$\begin{aligned} E_e^2 - (c|\vec{p}_e|)^2 &= m_e^2 c^4 \quad (|\vec{p}_e| \cos \phi)^2 \\ &= \left(\frac{hc}{\lambda} - \frac{hc}{\lambda'} + m_e c^2 \right)^2 - c^2 \left(\frac{h}{\lambda} - \frac{h}{\lambda'} \cos \phi \right)^2 \\ &\quad - c^2 \left(\frac{h}{\lambda'} \sin \phi \right)^2 \\ &\quad (|\vec{p}_e| \sin \phi)^2 \end{aligned}$$

Loads of cancellations:

$$\begin{aligned} m_e^2 c^4 &= \cancel{\left(\frac{hc}{\lambda} \right)^2} + \cancel{\left(\frac{hc}{\lambda'} \right)^2} + (m_e c^2)^2 \\ &\quad - 2 \frac{h^2 c^2}{\lambda \lambda'} + \cancel{2 \frac{hc}{\lambda} m_e c^2} - 2 \frac{hc}{\lambda'} m_e c^2 \\ &\quad - \cancel{\frac{h^2 c^2}{\lambda^2}} + \frac{2 h^2 c^2}{\lambda \lambda'} \cos \phi - \cancel{\frac{h^2 c^2}{\lambda'^2} \cos^2 \phi} \\ &\quad \quad \quad - \cancel{\frac{hc^2}{\lambda'^2} \sin^2 \phi} \end{aligned}$$

$$\lambda \lambda' \frac{hc}{\lambda \lambda'} (1 - \cos \phi) = hc m_e c^2 \left(\frac{1}{\lambda} - \frac{1}{\lambda'} \right) \times \lambda \lambda'$$

$$\boxed{(\lambda' - \lambda) = \frac{h}{m_e c} (1 - \cos \phi)}$$

$$\frac{h}{m_e c} \equiv \lambda_c = \text{Compton wavelength of electron}$$

First appearance... if you squeeze an electron down to smaller than this, an e^+e^- pop out!

$$= 2.43 \cdot 10^{-12} \text{ m}$$

$$= 2.43 \text{ pm}$$

$$E_\gamma = h\nu = \frac{hc}{\lambda_c} = \frac{hc}{\left(\frac{h}{m_e c}\right)} = m_e c^2 !$$

$$E_\gamma \cong 511,000 \text{ eV} = 0.511 \text{ MeV}$$

De Broglie Wavelength

particle momentum p

$$\lambda = \frac{h}{p}$$

way bigger

\Leftarrow

$$p = \frac{h}{\lambda}$$

photon!