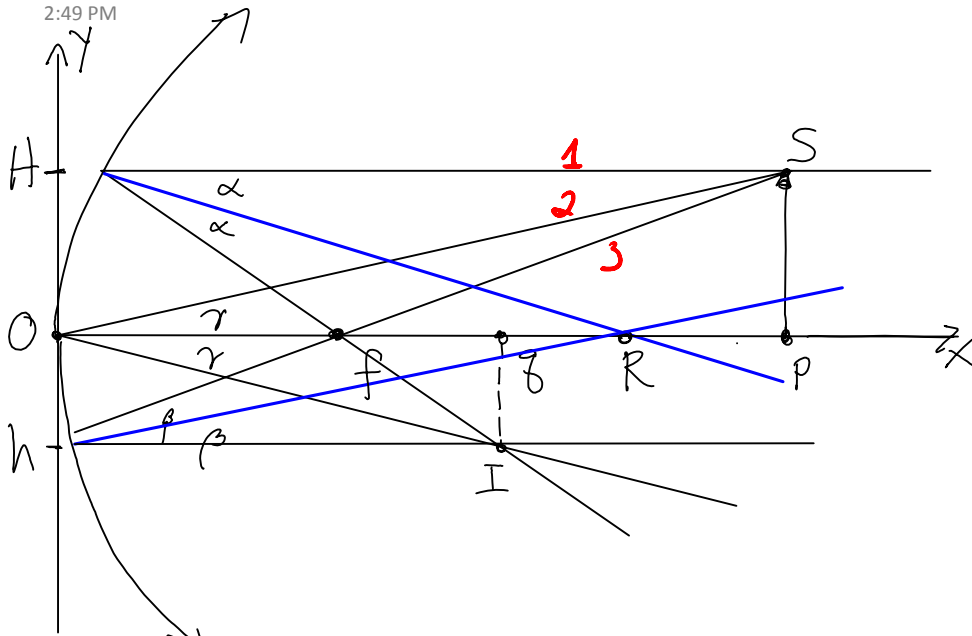


Nelson F. 1

Monday, May 26, 2008
2:49 PM



1st mark all distances & angles of the 3 principal rays and use the points S, I, O to mark the source, image & origin.

$$\tan \alpha = \frac{H}{R}, \tan(2\alpha) = \frac{H}{f} = \frac{2 \tan \alpha}{1 - \tan^2 \alpha}$$

$$y_{\alpha S}(x) = H$$

$$y_{\alpha R}(x) = \frac{H}{R}x + H$$

$$y_{\alpha I}(x) = -\frac{H}{f}x + H$$

$$\tan r = \frac{H}{p} = \frac{h}{q}$$

$$y_{r I}(x) = -\frac{H}{p}x = -\frac{h}{q}x$$

$$y_{\beta I}(x) = -h$$

$$\text{set } y_{\alpha I}(x) = -h = -\frac{H}{f}x + H$$

..

$$\text{set } Y_{\alpha I}(x_h) = -h = -\frac{H}{f} x_h + H$$

$$-h - H = -\frac{H}{f} x_h$$

$$\frac{fh}{H} + f = x_h$$

$$\text{set } Y_{\gamma I}(x_h) = -h = -\frac{H}{p} x_h$$

$$x_h = \frac{hp}{H}$$

for them to converge

$$\frac{fh}{H} + f = \frac{hp}{H}$$

$$\frac{h}{H} = \frac{2}{p}, \quad f = \frac{1}{2}R$$

$$\frac{R2}{2p} + \frac{1}{2}R = \frac{2p}{p}$$

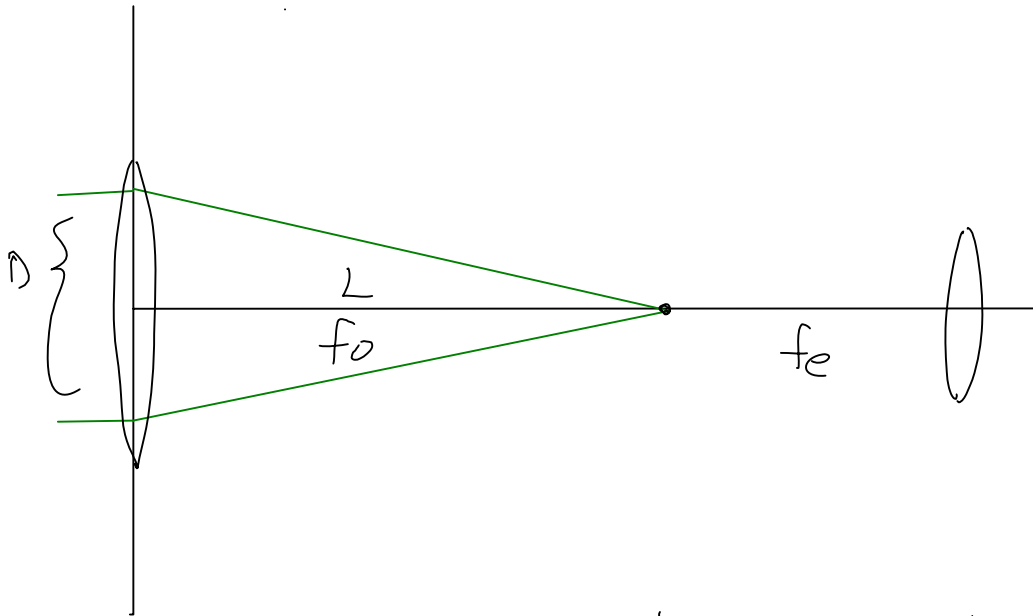
$$\frac{1}{2}R \left(1 + \frac{2}{p}\right) = \frac{2}{p}$$

$$\boxed{\frac{2}{R} = \frac{1}{2} + \frac{1}{p}}$$

Nelson 7.2

Monday, May 26, 2008

4:35 PM



The diffraction limit can be estimated by

$$\frac{D}{L} = \frac{\lambda}{d}$$

D = diameter of telescope

L = focal length

$\lambda \approx 500 \text{ nm}$.

$$d = \frac{500 \times 10^{-7} \text{ cm} \cdot 50 \text{ cm}}{5 \text{ cm}}$$

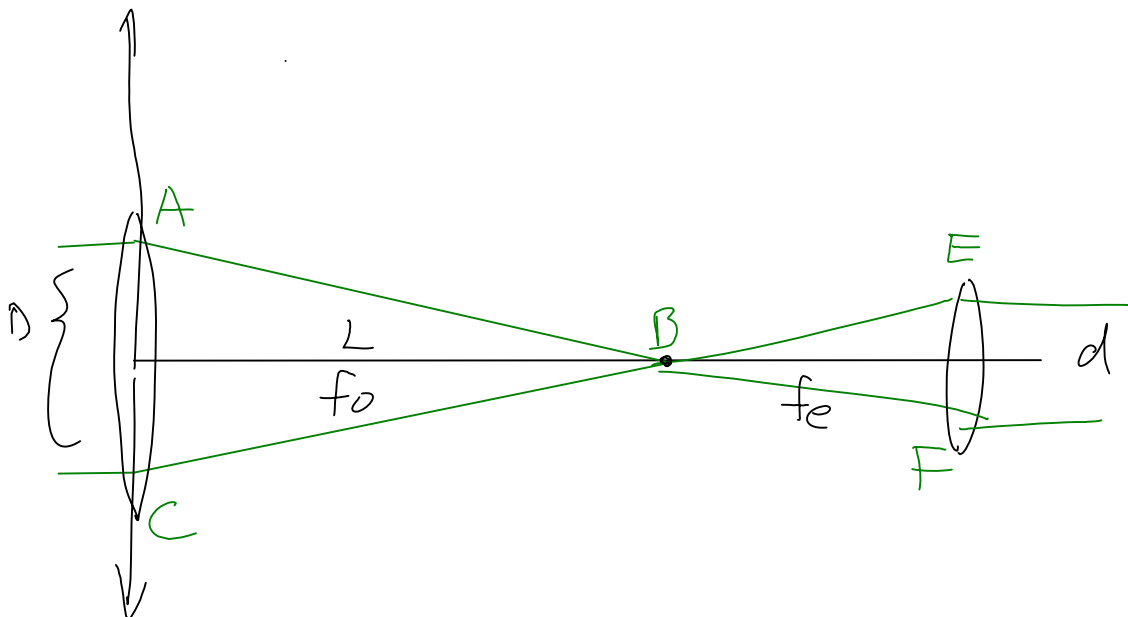
$$= 5 \times 10^{-4} \text{ cm}$$

$$M = \frac{f_o}{f_e} = 10 \rightarrow f_e = 5 \text{ cm}$$

Crawford 9.3a

Monday, May 26, 2008

4:49 PM



$$\triangle ABC \cong \triangle BEF$$

$$\frac{D}{f_o} = \frac{d}{f_e}$$

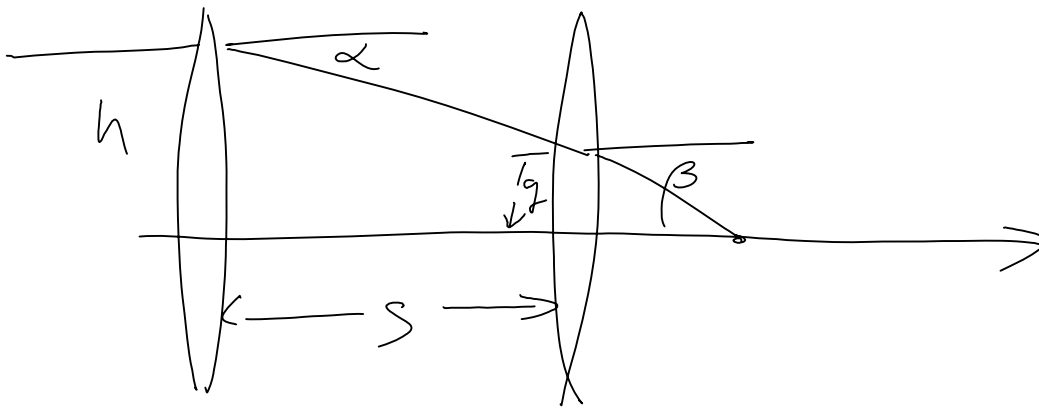
$$D = \frac{d f_o}{f_e}$$

d = diameter of pupil since all rays from telescope are parallel
Anything larger will not be seen by the naked eye.

Crawford 9.60

Monday, May 26, 2008

5:04 PM



$$\alpha \approx \frac{h}{f_1}$$

$$\beta \approx \frac{g}{f_2}$$

$$\frac{h}{F} = \tan(\alpha + \beta) \\ \approx (\alpha + \beta)$$

$$F = \frac{h}{\frac{h}{f_1} + \frac{g}{f_2}} = \frac{h}{\frac{h}{f_1} + \frac{h(1 - \frac{s}{f_1})}{f_2}}$$

$$g = h \left(1 - \frac{s}{f_1}\right)$$

$$F = \frac{1}{\frac{1}{f_1} + \frac{1}{f_2} - \frac{s}{f_1 f_2}}$$

$$F = \frac{1}{\frac{1}{f_1} + \frac{1}{f_2} - \frac{s}{f_1 f_2}}$$

for 1-convergent & 1-divergent

$$f_2 \rightarrow -f_2$$

$$F = \frac{1}{\frac{1}{f_1} - \frac{1}{f_2} + \frac{s}{f_1 f_2}}$$

if $f_2 = f_1$

$$F = \frac{f_1^2}{s}$$

Anderson 3.5 a

Monday, May 26, 2008

5:21 PM

positronium

e^+ e^-

muonium

p μ^-

Positronium

$$E = -\frac{\mu (Ze^2)^2}{2n^2\hbar^2}$$

$$\mu = \frac{m_0^2}{2m_0} = \frac{1}{2}m_0$$

$$E = -\frac{m_0}{4} \frac{e^4}{n^2\hbar^2}$$

$$= +\frac{1}{2} E_{\text{Hydrogen}}$$

$$r_n = 2 r_{n, \text{hydro}} = 1.06 \text{ \AA}$$

muonium

$$\mu = \frac{207m_0 + 1836m_0}{2043m_0}$$

$$\approx 186 m_0$$

$$E_n = 186 E_{\text{Hydrogen}}$$

$$\left(r_n = \frac{1}{186} r_{n \text{ Hydrogen}} \right)$$

Andersen 3.6

Monday, May 26, 2008

5:39 PM

1st excited state is $n=2$
transition to $n=1$ state

$$\text{Energy emitted} = W = E_2 - E_1$$

$$\mu = \frac{208 m_p m_{\alpha}}{208 m_p + m_{\alpha}}$$

$$= \frac{(208)(1836)(207) m_0}{m_0((208)(1836) + (207))}$$

$$= 206.89 m_0 \approx m_p$$

due to the fact that $m_{pb} \gg m_{\alpha}$.

$$\underline{Z_{pb} = 82}$$

$$E_n = + (82)^2 (207) (E_{N \text{ Hydrogen}})$$

$$W = \frac{3}{4} (E_1) = \frac{3}{4} (82)^2 (207) (13.6 \text{ eV})$$

$$\boxed{= 14.2 \text{ MeV}}$$

b) $W_{\text{meas}} = 5.963 \text{ MeV}$

$$b) \overline{W_{\text{meas}}} = 5.963 \text{ MeV}$$

i) size of nucleus

$$r_0 \sim \frac{0.53 \text{ \AA}}{(207)(82)} = 3 \text{ fm}$$

$$\text{radius of nucleus } r_A = A^{1/3} r_{\text{proton}}$$

$$r_{\text{proton}} = 1 \text{ fm}$$

$$r_{208} \sim 6 \text{ fm} > r_0$$

effective charge Z is less while the muon is in the nucleus.

ii) S & P orbitals are describe by different wave functions

P-orbitals are non-spherical which results in a diff. energy. See an undergraduate or google Hydrogen wave func.

iii) Pb_{208} -muon relativistic?

From eq. 3.24 of Anderson

$$\frac{1}{2} \mu v^2 = \frac{Z e^2}{2r}$$

for the hydrogen atom, let's try

for the hydrogen atom, let's try to find v .

$$v_n^2 = \frac{Ze^2}{r_n \mu}, \text{ where } n \text{ is the discrete states}$$

$$r_n = \frac{n^2 \hbar^2}{\mu Z e^2}$$

$$v_n^2 = \frac{Ze^2}{\frac{n^2 \hbar^2}{\mu Z e^2}}$$

$$= \frac{Z^2 e^4 \mu}{n^2 \hbar^2}$$

$$\alpha = \frac{e^2}{\hbar c}$$

$$v_n^2 = \frac{Z^2}{n^2} \alpha^2 c^2$$

$$\beta_n = \frac{v_n}{c} = \frac{Z}{n} \alpha$$

for hydrogen atom ground state

$$\beta = \alpha \leftarrow \text{easy way to remember Bohr atom.}$$

$$\bar{v}_e = \frac{1}{137} c \text{ non-rel.}$$

For Pb_{208} - muonic atom ground state

For Pb_{208} -muonic atom groundstate

$$\beta = \frac{f^2}{137} \alpha$$

$$\bar{v}_\mu \approx 0.6 C \rightarrow \text{relativistic}$$

classical treatment cannot be accurate at this "speed".